

NEPETIDONE AND NEPEDINOL, TWO NEW TRITERPENOIDS
FROM *NEPETA HINDOSTANA*

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Nepeta hindostana (Roth) Haines, (Labiatae) is an important medicinal plant of the Indo-Pakistan subcontinent. In the Greco-Arab system of medicine, popularly practised in Pakistan, the plant is known as "Badranj-e-Boya," and its decoction is used as a cardiac tonic, in fevers, and as a gargle for sore throat (1). Its alcoholic extract has been shown to possess hypocholesteremic activity (2).

We have reported earlier the isolation of new triterpenes of the lup-20(29)-ene series from this plant (3-6). We now record the isolation and structure elucidation of a new nortriterpenoid ketone, nepetidone (**1**), and a new triterpenoid tetraol, nepedinol (**2**), from the same source.

Nepetidone (**1**) was eluted with C_6H_6 -EtOAc (10:90) from the silica gel column, purified by repeated column chromatography and crystallized from MeOH as colorless crystals. It analyzed for $C_{29}H_{48}O_4$, and its uv spectrum in MeOH showed end absorption at 220 nm with a shoulder at 275 nm. The ir spectrum revealed the presence of hydroxyl ($3400-3200\text{ cm}^{-1}$ br), and ketone (1700 cm^{-1}). In the eims, the molecular ion peak is absent; the highest peak at m/z 442 represents the M^+-H_2O peak. However, the field desorption (fd) and the fabms show strong M^+ and M^++1 peaks at m/z 460 and 461, respectively. The eims shows further peaks at m/z 424, 370, 355, 327, 309, and 283 which are reminiscent of the ms of nepetidin, [1 β ,3 β ,11 α -trihydroxy-lup-20(29)-ene] (**3**), a triterpenoid isolated by us from the same plant (4). Some of the peaks of nepetidone are 2 mu higher than the corresponding ones of **3**, and the hrms shows that this is due to

the replacement of $C=CH_2$ of nepetidin with $C=O$ in nepetidone.

The relationship between nepetidin and nepetidone is clearly indicated by 1H - and ^{13}C -nmr spectroscopic studies. The 1H -nmr spectrum (300 MHz) of **1** in C_5D_5N shows methyl singlets at δ 0.78, 1.03 (6H, $2\times CH_3$), 1.11, 1.25, 1.32, and 2.10, the last singlet being due to the $COCH_3$ group. The spectrum further shows carbinyl proton signals at δ 3.59 (dd, $J=12$, 3.8 Hz, H-3), δ 3.98 (dd, $J=11$, 4.7 Hz, H-1) and δ 4.12 (m, H-11). There is a sextet centered at δ 2.67 ($J=11$, 11, 5.7 Hz) which is assigned to H-19 adjacent to a carbonyl group.

On acetylation, **1** yields a 3,11-diacetate (**1a**), as was observed also in the case of **3**. The 1 β -hydroxyl group, being sterically hindered, does not react with Ac_2O . In the 1H -nmr spectrum of the diacetate recorded in $CDCl_3$, the signals due to H-3 and H-11 are shifted to δ 4.50 (dd, $J=11.9$ Hz, 4.1 Hz), δ 4.94 (ddd, $J=9.5$, 9.5, 8.5 Hz), respectively, whereas the H-1 signal is seen as a dd at δ 3.68 ($J=10.9$, 4.98 Hz). The slight variation in the chemical shifts of H-1 in **1** and **1a** is due to the difference in solvent in which their spectra were recorded.

Table 1 shows the chemical shifts and assignments in the ^{13}C -nmr spectra of compounds **1** and **3** recorded in the same solvent (C_5D_5N). The ^{13}C -nmr spectrum of 3 β -hydroxy-30-norlupan-20-one (**4**) prepared from an authentic sample of lupeol by the OsO_4 method (7) is also included in Table 1 for comparison. The assignments were made on the basis of DEPT experiments as well as the known ^{13}C -nmr chemical shifts of lup-20(29)-ene derivative (3-6).

TABLE 1. ^{13}C -nmr Chemical Shifts (in ppm) of **1**, **2**, **3**, and **4** in $\text{C}_5\text{D}_5\text{N}$

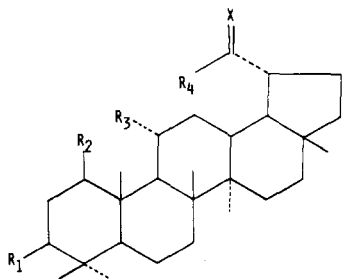
Carbon No.	Compound			
	1	2	3	4
1	66.45	66.69	66.57	39.29 ^m
2	35.29 ^a	37.65 ^c	36.24 ⁱ	27.88 ⁿ
3	75.06 ^b	75.15 ^f	75.12 ^j	78.10
4	39.96	42.82	39.97	39.53
5	58.23	58.25	58.22	55.88
6	18.78	18.82	18.80	18.81
7	35.29 ^a	35.80 ^e	35.85 ⁱ	34.66 ^o
8	42.70 ^c	43.00 ^g	42.74 ^k	41.08
9	52.19 ^d	53.52	53.50	50.70 ^p
10	46.33	46.40	46.34	37.34
11	76.50 ^b	76.61 ^f	76.54 ^j	21.15
12	34.96 ^a	35.12 ^e	35.09 ⁱ	27.56 ⁿ
13	35.23	36.31	36.19	37.53
14	42.76 ^c	43.27 ^g	43.10 ^k	42.41 ^q
15	28.05	32.41	30.20	27.71 ⁿ
16	38.01	38.06	37.95	35.30 ^o
17	43.16 ^c	43.27 ^g	43.17 ^k	43.22 ^q
18	49.42	49.09	48.73	49.66
19	53.45 ^d	44.01	48.19	52.51 ^p
20	211.36	156.27	150.60	211.38
21	27.84	28.06	28.0	28.31 ⁿ
22	39.96	40.04	40.10	40.18 ^m
23	29.39	28.68	28.67	29.23
24	14.22	14.38 ^h	14.37 ^l	16.18
25	15.65	15.70	15.66	16.36
26	17.67	17.75	17.75	16.42
27	14.22	14.28 ^h	14.25 ^l	14.67
28	18.23	18.06	18.31	18.13
29	—	106.64	110.27	—
30	28.64	64.44	19.39	28.68

^{a-p}Assignments may be reversed.

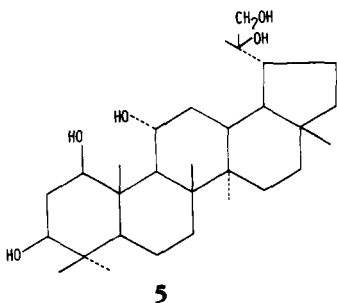
From the spectroscopic evidence, it is concluded that nepetidone has structure **1**. This is confirmed through chemical conversion of **3** into **1**; nepetidol (**3**) was treated with OsO_4 , and the pentaol (**5**) so formed was cleaved with periodic acid yielding **1**.

Nepedinol (**2**) was isolated from the fractions eluted with EtOAc-MeOH (95:5) from the silica gel column, purified by repeated column chromatography and crystallized from MeOH as small colorless crystals. Its fabms showed $\text{M}^+ + \text{H}$ peak at m/z 475, corresponding to the formula $\text{C}_{30}\text{H}_{50}\text{O}_4$. In the eims, no molecular ion peak is observed, the highest peak at m/z 456.3603 ($\text{C}_{30}\text{H}_{48}\text{O}_3$) corresponding to the $\text{M}^+ - \text{H}_2\text{O}$. It shows further peaks at

438, 384, 341, 323, 271, 201, 135, and 107. Some of the peaks of nepedinol are 16 mu higher than the corresponding ones of **3**. Its ir spectrum in KBr revealed the presence of hydroxyl (3350 cm^{-1}), but no band due to carbonyl group is observed. The uv spectrum had a maximum at 202 nm (end absorption). The $^1\text{H-nmr}$ spectrum ($\text{C}_5\text{D}_5\text{N}$, 300 MHz) of **2** showed six tertiary methyl signals at δ 0.84, 1.05 (s, 6H, $2 \times \text{CH}_3$), 1.13, 1.27, and 1.33. A singlet at δ 4.47 (2H) is due to CH_2OH , and nearby singlets with fine splitting at δ 5.12 and 5.41 can be assigned to the two H-29 protons. The secondary carbinyl proton signals are observed at δ 3.64 (m, H-3), δ 4.01 (dd, $J=11$, 4.7 Hz, H-1) and a distorted hextet at δ 4.13 (H-11). A



	R ₁	R ₂	R ₃	R ₄	X
1	OH	OH	OH	CH ₃	O
1a	OAc	OH	OAc	CH ₃	O
2	OH	OH	OH	CH ₂ OH	CH ₂
2a	OAc	OH	OAc	CH ₂ OH	CH ₂
3	OH	OH	OH	CH ₃	CH ₂
4	OH	H	H	CH ₃	O

**5**

broad signal at δ 4.47 is due to the two CH_2OH protons. Thus, it appears that nepedanol has a structure which is closely related to nepetidin (**3**). However, the absence of vinylic methyl, the presence of a CH_2OH group, only six tertiary methyl groups, and the relative downfield shift of H-29 signal, all indicate that C-30 contains a primary hydroxyl group.

All of the spectroscopic data cited above indicate that the structure of nepedanol is $1\beta,3\beta,11\alpha,30$ -tetrahydroxy-lup-20(29)-ene. This proposed structure was also supported by the ^{13}C -nmr spectrum of **2** (Table 1) which shows that there are four carbon atoms bearing oxygen functions at δ 66.69 (C-1), 75.15 (C-3), 76.61 (C-11), and 64.44 ppm (C-30).

Acetylation of **2** with Ac_2O and pyridine furnished at 3, 11, 30-triacetate (**2a**). In the ^1H -nmr spectrum of the triacetate recorded in CDCl_3 , the signal

due to H-30 is shifted to δ 4.52 (brs) partly superimposed by a double doublet centered at δ 4.51 due to H-3. The H-11 multiplet at δ 4.95 is masked by two broad singlets at δ 4.90 and δ 4.97 due to the two H-29 protons. The H-1 multiplet is observed at δ 3.64.

On the basis of the above spectra, structure **2** is suggested for nepedanol. This structure was also supported by a comparison of the ^{13}C -nmr spectral data of **2** with those of **1** and **3**. (Table 1).

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—Melting points were determined on a Gallenkamp melting point apparatus. Uv spectra were measured in MeOH with a Shimadzu UC 240 Graphicord spectrometer. Ir spectra were prepared with a Jasco A-302 spectrometer. The ^1H -nmr (300.13 MHz) and ^{13}C -nmr (75.43 MHz) spectra were recorded on a Bruker AM-300 spectrometer in $\text{C}_5\text{D}_5\text{N}$. The DEPT experiments were carried out with $\theta=45^\circ, 90^\circ$, and 135° ; the quaternary carbons were determined by subtraction of these spectra from the broad band ^{13}C -nmr spectrum. The ei, fd, and fabms spectra were recorded on a Finnigan MAT 312 double focusing ms spectrometer coupled with PDP 11/34 computer system. Tlc was carried out on silica gel plates using the following solvent systems: CHCl_3 -MeOH (17:3), (19:1); (8:2), and (9:1); spots were visualized by spraying with ceric sulfate solution in 10% H_2SO_4 followed by heating.

PLANT MATERIAL.—The plant material was purchased from the local market and identified by the Department of Pharmacognosy, University of Karachi. A voucher specimen has been deposited in the herbarium of the Department of Botany, University of Karachi.

EXTRACTION AND ISOLATION.—The plant material was extracted three times with *n*-hexane and then with EtOH. The combined ethanolic extract was evaporated under reduced pressure, leaving behind a greenish, syrupy residue which was then partitioned between EtOAc and H_2O . The EtOAc layer was evaporated under reduced pressure, yielding a green, gummy material which was then subjected to column chromatography on silica gel. Elution was carried out with a gradient of increasing polarity in the order of *n*-hexane, C_6H_6 , EtOAc, and MeOH.

NEPETIDONE (1).—Compound **1** was eluted from the silica gel column with C_6H_6 -EtOAc (10:90). It was further purified by repeated (three times) column chromatography on silica gel, and

crystallized from MeOH as colorless crystals (25 mg); mp 300° dec; $[\alpha]_D -28.13^\circ$ ($c=0.32$, MeOH); uv λ max (MeOH) 220 nm (end absorption); ir ν max (KBr) 3200-3500 br (OH), 1700 (C=O), 1460, 1382, 1360, 1180, 1000 cm^{-1} ; ^1H nmr (300 MHz, $\text{C}_5\text{D}_5\text{N}$) δ 0.78 (s, CH_3), 1.03 (s, 6H, $2\times\text{CH}_3$), 1.11 (s, CH_3), 1.25 (s, CH_3), 1.32 (s, CH_3), 2.10 (s, COCH_3), 2.67 (sext, $J=11$, 11, 5.7 Hz, H-19), 3.59 (dd, $J=12$, 3.8 Hz, H-3), 3.98 (dd, $J=11$, 4.7 Hz, H-1), 4.12 (m, H-11); ^{13}C nmr see Table 1; fdms 460 (M^+); fabms 461 (M^++1); eims M^+ absent, 442.344 (6, $\text{M}^+-\text{H}_2\text{O}$, calcd for $\text{C}_{29}\text{H}_{46}\text{O}_3$, 442.345), 424 (12, $\text{M}^+-2\text{H}_2\text{O}$), 370.322 (31, calcd. for $\text{C}_{26}\text{H}_{42}\text{O}$, 370.323), 327.268 (88, calcd for $\text{C}_{23}\text{H}_{35}\text{O}$, 327.269), 309.257 (26, calcd. for $\text{C}_{23}\text{H}_{33}$, 309.258), 257 (8), 283 (12), 231 (14), 205 (30), 163 (34), 135 (84), 107 (100), 95 (88). *Anal.* calcd for: $\text{C}_{29}\text{H}_{48}\text{O}_4$. C, 75.60; H, 10.50. Found: C, 73.67; H, 10.57.

ACETYLATION OF 1.—Compound 1 (10 mg) was dissolved in pyridine (1 ml) and treated with Ac_2O (3 ml) at room temperature overnight. Ice was added to the reaction mixture which was extracted with EtOAc and H_2O . The EtOAc layer was evaporated to yield the diacetate (1a), which could not be crystallized. It, however, appeared pure by tlc. Ir ν max (KBr) 3500-3400 br (OH), 2850, 1750 (OCOCH_3), 1300, 1260-1160, 1000 cm^{-1} ; ^1H nmr (300 MHz, CDCl_3) δ 0.76 (s, CH_3), 0.82 (s, CH_3), 0.85 (s, CH_3), 0.97 (s, CH_3), 1.00 (s, 6H, $2\times\text{CH}_3$), 1.94 (s, OCOCH_3), 2.04 (s, OCOCH_3), 2.14 (s, COCH_3), 2.61 (sext, $J=11$, 11, 5.67 Hz, H-19), 3.68 (dd, $J=10.9$, 4.98 Hz, H-1), 4.50 (dd, $J=11.9$, 4.1 Hz, H-3), 4.94 (ddd, $J=9.5$, 9.5, 8.5 Hz, H-11); eims m/z M^+ absent, 484 (2, M^+-AcOH), 466 (4, $\text{M}^+-\text{AcOH}-\text{H}_2\text{O}$), 424 (6, M^+-2AcOH), 406 (8, $\text{M}^+-2\text{AcOH}-\text{H}_2\text{O}$), 327 (100), 309 (26), 283 (12), 231 (14), 205 (30), 163 (34), 135 (84), 107 (100), 95 (68).

CONVERSION OF 3 INTO 1.—Compound 3 was treated with OsO_4 (8) in the presence of dioxane for 5 days to yield the pentaol (5); eims m/z M^+ absent, 456 (3, $\text{M}^+-2\text{H}_2\text{O}$), 438 (6), 384 (4), 341 (18), 283 (28), 161 (38), 135 (62), 107 (100), 95 (98). 5 was cleaved with periodic acid yielding 1, which was identified through c-tlc, superimposable ir spectra and mixed melting points.

NEPEDINOL (2).—Compound 2 was eluted from the fractions eluted with CHCl_3 -MeOH

(19:1), purified and crystallized as described above in the case of 1; mp 282° dec.; $[\alpha]_D -18.67^\circ$ ($c=0.75$, $\text{C}_5\text{D}_5\text{N}$); uv λ max (MeOH) 202 nm (end absorption); ir ν max (KBr) 3350 (OH), 2950 cm^{-1} ; ^1H nmr (300 MHz, $\text{C}_5\text{D}_5\text{N}$) 0.84 (s, CH_3), 1.05 (s, 6H, $2\times\text{CH}_3$), 1.13 (s, CH_3), 1.27 (s, CH_3), 1.33 (s, CH_3), 3.64 (m, H-3), 4.01 (dd, $J=11$, 4.7 Hz, H-1), 4.13 (distorted hext, H-11), 4.47 (s, CH_2OH), 5.12 (brs) and 5.41 (brs) ($2\times\text{H}-29$); ^{13}C nmr see Table 1; fabms 475 (M^++1); eims m/z (M^+ absent), 456.361 (6, $\text{M}^+-\text{H}_2\text{O}$, calcd. for $\text{C}_{30}\text{H}_{48}\text{O}_3$, 456.360), 438.351 (8, $\text{M}^+-2\text{H}_2\text{O}$, calcd. for $\text{C}_{30}\text{H}_{46}\text{O}_2$, 438.349), 384.338 (12, calcd. for $\text{C}_{27}\text{H}_{44}\text{O}$, 384.339), 341.285 (32, calcd. for $\text{C}_{24}\text{H}_{37}\text{O}$, 371.284), 323.274 (34, calcd. for $\text{C}_{24}\text{H}_{35}$, 323.273), 271 (6), 201 (22), 135 (60), 107 (95), 95 (100).

ACETYLATION OF 2.—Compound 2 was acetylated as 1 to yield 2a. Ir ν max (CDCl_3) 3500 (OH), 2950, 1700 (OCOCH_3), 1240 (C-O stretching), 1020, 760 cm^{-1} ; ^1H nmr (300 MHz, CDCl_3) δ 0.77 (s, CH_3), 0.82 (s, CH_3), 0.85 (s, CH_3), 0.97 (s, CH_3), 0.98 (s, CH_3), 1.02 (s, CH_3), 1.98 (s, OCOCH_3), 2.04 (s, OCOCH_3), 2.09 (s, OCOCH_3), 3.64 (m, H-1), 4.51 (dd, H-3), 4.52 (br.s, H-30), 4.90 and 4.97 (br.s, $2\times\text{H}-29$); eims m/z M^+ absent, 540 (4, M^+-AcOH), 480 (6, M^+-2AcOH), 462 (3, $\text{M}^+-2\text{AcOH}-\text{H}_2\text{O}$), 420 (4), 383 (31), 323 (100), 267 (7), 201 (28), 107 (80), 95 (74).

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